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A Convenient Synthesis of'Optically Pure (2R, 3R)-2,3- Epoxysuccinyl - Dipeptides

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Abstract: (2R,3R)-trans-Epoxysuccinyl-dipeptides (7a-7d) were synthesized by acylation of dipeptides with the Nhydroxysuccinimde or pentafluorophenyl ester of monoethyl (2R,3R)-trans-epoxysuccinate (6a-6b). A nucleophilic oxirane ring opening by N-hydroxysuccinimide and pentafluorophenol during the preparation of the active esters could not be **observed. Subsequent saponification of the monoethyl ester 7a-7d with KOH in ethanolic solution allowed to produce the dipeptide derivatives as potassium salts (8a-8d) which were found to be stable on storage in the cold. The attempts to convert** (2R,3R)-trans-epoxysuccinyl-glycyl-proline (7a) into the corresponding (2S,3S)-trans-epithiosuccinyl derivative via an oxygen **sulfur exchange reaction with 3-methylbenzothiazole-2-thione failed completely as among the various prodocts of onknown** nature formed only the desulfurated fumaryl derivative could be isolated and characterized.

Introduction

Since the isolation of L-trans-epoxysuccinyl-leucylamido(4-guanidino)butane (E-64) from extracts of Aspergillus Japonicus TPR-46 and its discovery as a potent new inhibitor of cysteine-proteases¹, increasing interest has been paid to the epoxysuccinyl moiety as reactive handle for the design of specific inhibitors of this class of proteases². Because of the fast reaction with the active-site thiol function^{2c,2d,3}, the high degree of specificity for cysteine proteases^{1,2c} and the low toxicity⁴ epoxysuccinyl-peptide derivatives represent a promising new type of therapeutic agents as cysteine proteases are known to be implicated in a variety of disease states. Nevertheless surprisingly little work has been reported on the optimization of synthetic protocols.

Although the D -trans-epoxysuccinyl-peptide derivatives are known to be weaker inhibitors than the related L-isomers we have focussed our attention on the synthesis of **this** isomer in view of a possible direct conversion of the epoxysuccinyl-peptides into the episulfides by oxygen sulfur exchange reactions⁵ which occur with Walden inversion at both chiral C-atoms. This approach would allow us to investigate possible improvements in the synthetic accessibility of epoxysuccinyl-peptides independently of the *D-* or Lconfiguration and finally to attempt their conversion into the L -trans-epithiosuccinyl analogs as possibly interesting new inhibitors of cysteine-proteases. In Eact, the thiirane moiety is expected to be at least as reactive as the oxirane ring and to form in analogy to the epoxy group a stable thioetber bond with the active-site thiol function⁶.

Results and Discussion

The key intermediate in the synthesis of D - (or L -) *trans*-epoxysuccinyl-peptides^{*} is the related monoalkylepoxysuccinate. This derivative is available by resolution of racemic epoxysuccinic acid via diastereomeric salt formation⁷ followed by esterification and partial saponification⁸ or by stereoselective synthesis of the dialkyl ester of desired configuration⁹ and partial hydrolysis to the monoalkyl ester. We have selected the second approach^{9a} in a slightly modified version which allowed us to prepare from diethyl L -(+)-tartrate **(1)** following the route outlined in Sheme 1 the diethyl D -trans-epoxysuccinate (4) as optically homogeneous compound.

Saponification of 4 with the stoichiometric amount of KOH produced the desired monoethyl *D-trms*epoxysuccinate potassium salt **(5a)** in a satisfactory yield of about 49% over the three steps. This compound proved to be fully stable on storage in the cold over longer periods of time. A crucial step in this synthesis was found to be, upon fractional distillation of the diethylester 4, a silica gel chromatography with

^{*}Abbreviations: Standard abbreviations as recommended by the IUPAC-IUB Commission on Biochemical Nomenclature (J. Biol. Chem. 1972, 247, 977) are used for amino acids and related derivatives; Agm, agmatine; DCC, dicyclohexylcarbodiimide; DMF, dimethylformamide; DMSO; dimethyl sulfoxide; EtOAc, ethyl acetate; EtOH, ethanol; HOAc, acetic acid; HOBt, 1-hydroxybenzotriazole; MBTT, 3-methylbenzothiazole-2-thione; MeOH, methanol; NMM, Nmethylmorpholine; tEps, (2R,3R) trans-epoxysuccinyl; TFA, trifluoroacetic acid.

petroleum ether/EtOAc as eluent. This allowed for separation of trace amounts of contaminating compound 2 and 3 as monitored by MS and NMR.

N^o-acylation of suitably protected amino acid or peptide derivatives with monoalkyl-epoxysuccinate has been performed, so far, using diethylphosphorylcyanide^{1b} or DCC/HOBt^{2e,2f,8} as coupling reagents as well as by the 4-nitrophenyl ester procedure'a. Thereby relatively low yields of epoxysuccinyl-peptides were obtained, a fact which could possibly be attributed to sterical hindrance from the oxirane ring or to side reactions involving nucleophilic ring opening, e.g. by HOBt. We have analyzed both aspects by the use of the N-hydroxysuccinimide **(6a)** and pentatluorophenyl ester **(6b)** which are known to be more reactive than the 4-nitrophenyl ester. Moreover N-hydroxysuccinimide and pentatluorophenol are expected to be weaker nucleophiles than HOBt and 4-nitrophenol, respectively. Although the epoxysuccinyl moiety is known to undergo facile attack by nucleophiles, this type of side reaction could not be observed during the preparation of the active esters using equimolar amounts of N-hydroxysuccinimide as a good nucleophile and pentafluorophenol as a weak nucleophile. With or without intermediate isolation of the active esters unprotected dipeptides were acylated with more or less success in terms of final yields as reported in Scheme 2.

The N-hydroxysuccinimide ester procedure allowed to significantly enhance the yield of E-64 as D -isomer **(8d), upon saponification of the monoethyl ester 7d with KOH, from 8%^{1b} to 23% in the overall reaction.** Similarly, by the pentatluorophenyl ester method compound **8b wae** obtained over the two steps in 26% **yield in comparison to** the 9% yield reported for this epoxysuccinyl-peptide by the DCC/HOBt procedure2e. The monoethyl esters 7a-d were converted to the free acids (8a-d) by saponification with KOH as enzymatic hydrolysis with pig liver esterase has been found to produce the free acids in relatively low $yields^{2e}$ (Scheme 3).

Scheme 3

If needed, e.g. for 8c and 8d, preparative HPLC was used for the final purification whereby attention was paid to avoid the use of acids and buffer ingredients since nucleophilic oxirane ring opening was observed to occur particularly, in the lyophilization step.

Finally, the D-trans-epoxysuccinyl-glycyl-proline monoethyl ester (7a) was treated as model compound with 3-methylbenzothiazole-2-thione which has been proposed as the most efficient reagent for an oxygen sulfur exchange in the oxirane ring^{5e}. Following the literature procedure, i.e. operating at 0 °C, as well as by refluxing the reaction mixture in CH_2Cl_2 as described for similar reactions with phosphine sulfides^{5d} all attempts to produce the corresponding L-trans-epithiosuccinyl analog failed (Scheme 4).

A complex mixture of products was formed of which none of the components was found to exhibit the expected mass. Among the various products isolated one was identified as the desuffirated derivative according to the mass spectrum. Therefore alternative procedures are under investigation to possibly produce these epithiosuccinyl-peptides.

Experimental Procedures

Melting points were determined on a capillary melting point apparatus (Büchi) and are uncorrected. Elemental analysis were obtained with a Heraeus CHN-O-Rapid apparatus. Optical rotations were measured in a 1 dm cell on a Perkin Elmer polarimeter (model 241). 1D and 2D ¹H- as well as ¹³C-NMR spectra were recorded at 300 K using a Bruker AM 500 spectrometer at 500 MHz for ¹H and a Bruker AM 400 spectrometer at 100.6 MHz for ¹³C {¹H}, respectively. FAB-MS spectra were obtained on a Finnigan MAT 900 and EI-MS spectra on a Varian MAT 312, respectively. FT-IR data were determined with a Perkin Elmer FT-IR Spectrometer 1760 X. Analytical HPLC was performed on either Nucleosil C₈ (4 x 25 cm, 5) μ m particle size, 300 Å) or C₁₈ (4 x 25 cm, 5 μ m particle size, 100 Å) columns (Macherey-Nagel, Düren, Germany) using the following buffers: A) 5% CH₃CN/95% H₃PO₄ (2%); B) 15% CH₃CN/85% H₃PO₄ (2%); C) 50% CH₃CN/50% H₃PO₄ (2%); flow rate: 1 mL/min. Preparative HPLC was performed on a Macherey-Nagel Nucleosil C₁₈ (2 x 25 cm, 5 µm particle size, 300 Å) column using following buffers: I) 2% CH₃CN/98% H₂O; II) 5% CH₃CN/95% H₂O; III) 15% CH₃CN/85% H₂O; IV) 50% CH₃CN/50% $H₂O$; flow rate: 9 mL/min.

The following dipeptides were obtained by standard procedures of peptide synthesis: **H-Gly-Pro-OH**¹⁰, **HCOOH-H-Leu-Agm-1/2** H_2SO_4 (semi solid; FAB-MS: m/e = 244.5 (100%), M+H⁺ = 244.4 calcd. for $C_{11}H_{25}N_5O$); **H-Leu-Arg-OH**¹¹; **H-Leu-Pro-OH**¹¹.

Diethyl (ZR, 3S)-erythro-2-acetosy-3-bromosuccinate (2)

Compound 2 was prepared from l(85.6 **g, 0.415** mol) by treatment with **33% HBr/HOAc (292 mL) for 4h** at room temperature; yield 2: 106 g (82%); FT-IR: v_{max} = 2985 (m), 1756 (s), 1374 (m), 1275 (m), 1212 (s), 1024 (m), 860 (m) cm⁻¹; ¹H-NMR (CDCl₃): δ = 1.30, 1.31 (2 x t, 6H, J = 7 Hz, 2 x <u>CH₃</u>-CH₂-O-); 2.17 (s, 3H, CH₃-CO); 4.24, 4.28 (2 x q, 4H, 2 x CH₃-CH₂-O-); 4.81 (d, 1H, J = 5 Hz, C3-H); 5.59 (d, 1H, $J = 5$ Hz, C2-H); anal. calcd. for $C_{10}H_{15}O_6Br$: C, 38.60; H, 4.86, found: C, 38.45; H, 5.02.

Diethyl (2R, 3S)-erythro-2-hydroxy-3-bromosuccinate (3)

The intermediate 2 (106 g, 0.339 mol) was treated with 32 mL of 33% HBr/HOAc in EtOH (962 mL) under reflux for 5h. Fractional distillation without further silica gel chromatography led to the homogeneous 3; yield: 84.1 g (92%); bp 111-112 °C/~0.2 mm (Lit.^{9a}: 123-125 °C/0.6 mm); n_D²³ = 1.4635 (Lit.^{9a}: n_D²¹ = 1.4628); $[\alpha]_D^{23} = -29.4$ ° [neat] (Lit.^{9a}: $[\alpha]_D^{21} = -28.9$ °); FT-IR: $v_{max} = 3481$ (m), 2984 (m), 1742 (s), 1468 (w), 1447 (w), 1371 (m), 1267 (s), 1222 (s), 1160 (s), 1114 (m), 1024 (s), 861 (m) cm-l; IH-NMR $(CDC1₃)$: $\delta = 1.01$, 1.03 (2 x t, 6H, J = 7 Hz, 2 x CH₃-CH₂-O-); 3.94-4.03 (m, 5H, 2 x CH₃-CH₂-O-, -OH); 4.41, 4.43 (2 x d, 2H, $J_1 = 5$ Hz, C3-H, $J_2 = 5$ Hz, C2-H).

Diethyl(2R,3R)-trans-2,3-Epoxysuccinate (4)

Compound 4 was obtained from 3 (73.7 g, 0.274 mol) in EtOH (63 mL) by reaction with 7.56 g (0.329 mol) of sodium in EtOH (96 mL) for lh at room temperature. After fractional distillation the product was chromatographed in aliquots of 1.5 g on a silica gel column $(3 \times 45 \text{ cm})$ using petroleum ether/EtOAc $(4:1)$ as eluent. The product was isolated as colourless oil in 48% yield; bp 80-82 °C/~0.2 mm (Lit. $9a:100-104$ $^{\circ}$ C/4 mm); n_D²² = 1.4384 (Lit.^{9a}: n_D²³ = 1.4354); [α]_D²³ = -102.7 $^{\circ}$ [c = 1 ether] (Lit.^{9a}: [α]_D^{21.5} = -88.5 "); HPLC (Nucleosil C₈): linear gradient from 100% B to 90% C in 25 min, t_R 12.59 min; FT-IR: $v_{\text{max}} =$ 2986 (m), 1748 (s), 1469 (w), 1448 (w), 1372 (m), 1330 (m), 1280 (m), 1249 (m), 1231 (m), 1200 (s), 1096 (m), 1029 (s), 964 (w), 901 (m), 859 (w), 841 (w) cm⁻¹; ¹H-NMR (CDCl₃): δ = 1.26 (2 x t, 6H, J = 7.2 Hz, 2 x CH_3 -CH₂-O-); 3.61 (s, 2H, Eps C-H, C-H); 4.21 (q, 4H, J = 7.1 Hz, 2 x CH₃-CH₂-O-); ¹H-NMR (DMSO): $\delta = 1.23$ (2 x t, 6H, J = 7.2 Hz, 2 x CH₃-CH₂-O-); 3.75 (s, 2H, Eps C-H, C-H); 4.19 (q, 4H, J = 7.1 Hz, 2 x CH₃-CH₂-O-); ¹³C-NMR (DMSO): δ = 13.8 (2 x CH₃-CH₂-O-); 51.5 (Eps C-H, C-H); 61.6 (2 x CH₃-CH₂-O-); 166.4 (2 x CO).

Ethyl (2R, 3R)-irens-2,3-Epoxysuccinate Potassium Salt (5a)

To an ice-cold solution of 4 (1.2 g, 6.4 mmol) in EtOH (5 mL) an ethanolic solution (11 mL) of KOH (359 mg, 6.4 mmol) was added dropwise. After additional 2h stirring at 0 "C the reaction mixture was diluted with ether, and the precipitate was collected by filtration and dried under reduced pressure at 30 °C to give

a colourless powder; yield 5a: 1.0 g (79%); mp >159 °C decomp.; FAB-MS: m/e = 159.4 (100%), M⁺ = 159.1 calcd. for $C_6H_7O_5$; HPLC (Nucleosil C_8): linear gradient from 100% B to 40% C in 25 min, t_R 4.65 min (98%); ¹H-NMR (D₂O): $\delta = 1.25$ (t, 3H, J = 7.0 Hz, CH₃-CH₂-O-); 3.49 (d, 1H, J = 2.0 Hz, Eps C-H); 3.58 (d, 1H, J = 2.0 Hz, Eps C-H); 4.25 (q, 2H, J = 7.0 Hz, CH₃-CH₂-O-).

Ethyl (2R, 3R)-trams-2,3-Epoxysuccinate (5h)

To a solution of 5a (512 mg; 2.58 mmol) in 5% KHSO₄ (50 mL) NaCl was added and the free acid was extracted with EtOAc ($3x50$ mL). The combined organic layers were washed with water ($1x50$ mL), dried over Na₂SO₄ and then evaporated to a colourless oil; yield 5b: 320 mg (77%); EI-MS: m/e = 161 (2%), 87 (100%), 115 (50%), M+H⁺ = 161.1 calcd. for $C_6H_8O_5$.

N-[[(2R,3R)-3-trans-(Ethoxycarbonyl)oxiran-2-yl]carbonyl]-glyeyl-proline (7a)

5b (320 mg, **2.0** mmol) in DMF (2 mL) was reacted with pentatluorophenol(368 mg, 2.0 mmol) and DCC (412 mg, 2.0 mmol) in DMF (each in 4 mL) at 0 °C. The reaction mixture was allowed to reach room temperature and was then stirred for additional 24h. Dicyclohexylurea was filtered off and the filtrate was evaporated to dryness under reduced pressure. H-Gly-Pro-OH (344 mg, 2.0 mmol) in DMF (15 mL) was added to the residue and and the reaction was allowed to proceed for 24h at room temperature. Then the solvent was evaporated and the crude reaction mixture was chromatographed on a silica gel column (3 x 45) cm) using CHCl₃/MeOH (1:1) as eluent. Fractions containing homogeneous product were collected, concentrated to small volume and the product was precipitated with ether; yield **7a:** 346 mg; (55%); mp >195 °C decomp.; FAB-MS: m/e = 315.0 (50%), M+H⁺ = 315.3 calcd. for C₁₃H₁₈N₂O₇; HPLC (Nucleosil C_{18}): linear gradient from 100% A to 40% C in 25 min, t_R 11.66 min (96%); ¹H-NMR (DMSO): $\delta = 1.23$ (t, 3H, J = 7.1 Hz, CH_3 -CH₂-O-); 1.67-1.72 (m, 2H, β CH₂ Pro); 2.01-2.05 (m, 2H, γ CH₂ Pro); 3.29-3.37 (m, 4H, δ CH₂ Pro, α CH₂ Gly); 3.60 (d, 1H, J = 1.7 Hz, Eps C-H); 3.79 (d, 1H, J = 1.6 Hz, Eps C-H); 3.96 (dd, 1H, J = 5.8 Hz, J = 16.8 Hz, α CH Pro); 4.19 (q, 2H, J = 7.0 Hz, CH₃-CH₂-O-); 8.52 (br s, 1H, NH Gly); ¹³C-NMR (DMSO): δ = 13.9 (CH₃-CH₂-O-); 22.1 (γ CH₂ Pro cis); 24.6 (γ CH₂ Pro trans); 28.9 (β CH₂ Pro trans); 31.2 (β CH₂ Pro cis); 41.3 (α CH₂ Gly); 45.5 (δ CH₂ Pro trans); 46.2 (δ CH₂ Pro cis); 51.2 (Eps C-H); 52.8 (Eps C-H); 60.8 (CH₃-CH₂-O-); 61.0 (α CH Pro trans); 61.4 (α CH Pro cis); 164.9, 164.9, 166.4 (3 x CO); 167.1 (-COOH); ¹³C-NMR (D₂O): δ = 16.0 (CH₃-CH₂-O-); 25.1 (γ CH₂ Pro cis); 27.1 (γ CH₂ Pro trans); 32.3 (β CH₂ Pro trans); 34.5 (β CH₂ Pro cis); 44.3 (α CH₂ Gly); 49.6 (δ CH₂ Pro trans); 50.2 (δ CH₂ Pro cis); 55.4 (Eps C-H); 56.3 (Eps C-H); 64.7 (CH₃-CH₂-O-); 64.9 (α CH Pro cis); 66.2 (a CH Pro trans); 170.6, 171.3, 171.7 (3 x CO); 182.5 (-COOH).

N-[[(2R,3R)-3-trans-(Ethoxycarbonyl)oxiran-2-yl]carbonyl]-leucyl-proline (7b)

Compound 7b was synthesized as described for 7a by acylation of H-Leu-Pro-OH with freshly prepared 6b. The crude reaction product was partitioned between EtOAc and 5% KHSO₄ (3 x), washed with water (2 x) and dried over $Na₂SO₄$. The combined organic layers were evaporated and the residue was chromatographed on silica gel using $CHCl₃/MeOH$ (2:1) as eluent and finally purified by preparative HPLC using an isocratic eluent of 90% III and 10% IV; fractions containing homogeneous material were combined and lyophilized twice from water; yield **7b:** 28%; mp >155-160 'C decomp.; FAB-MS: m/e = 370.7 (26%), $M^+=$ 370.4 calcd. for $C_{17}H_{26}N_2O_7$; m/e = 393.3 (30%), $M+Na^+=$ 393.4; m/e = 409.5 (26%), $M+K^+=$ 409.5; HPLC (Nucleosil C₈): linear gradient from 100% B to 40% C in 20 min, t_p 20.76 min (98%); ¹H-NMR (DMSO): $\delta = 0.85$ (d, 3H, J = 6.6 Hz, δ_2 CH₃ Leu); 0.89 (d, 3H, J = 6.4 Hz, δ_1 CH₃ Leu); 1.22 (t, 3H, J = 6.9 Hz, CH₃-CH₂-O-); 1.44-1.54 (m, 3H, β CH₂ Leu, γ CH Leu); 1.62-1.65 (m, 1H, β ₂ CH₂ Pro); 1.83-1.95 (m, 3H, γ CH₂ Pro, β_1 CH₂ Pro); 3.47-3.49 (m, 1H, δ_2 CH₂ Pro); 3.57-3.62 (m, 1H, δ_1 CH₂ Pro); 3.59 (d, 1H, J = 1.8 Hz, Eps C-H); 3.74 (d, 1H, J = 1.8 Hz, Eps C-H); 4.18 (q, 2H, J = 6.9 Hz, CH₃-CH₂-O-); 4.19-4.22 (m, 1H, α CH Pro); 4.57 (m, 1H, α CH Leu); 8.74 (d, 1H, J = 8.2 Hz, NH Leu); ¹³C-NMR (DMSO): $\delta = 13.9$ (CH₃-CH₂-O-); 21.5 (δ CH₃ Leu); 21.9 (γ CH₂ Pro cis); 23.2 (δ CH₃ Leu); 24.0 (γ CH Leu); 24.2 (γ CH₂ Pro trans); 28.9 (β CH₂ Pro trans); 31.1 (β CH₂ Pro cis); 40.1 (β CH₂ Leu); 46.1 (δ CH₂ Pro); 48.8 (α CH Leu); 51.0 (Eps C-H); 52.7 (Eps C-H); 60.6 (CH₃-CH₂-O-); 61.4 (α CH Pro); 164.5, 167.2, 168.7 (3 x CO); 169.7 (-COOH).

N-[[(2R,3R)-3-trans-(Ethoxycarbonyl)oxiran-2-yl]carbonyl]-leucyl-arginine (7c)

The N-hydroxysuccinimide ester **6a** was prepared as in DMF from **5b,** N-hydroxysuccinimide and DCC at 1: 1: 1 molar ratio and then reacted with H-Leu-Arg-OH as described for **7a.** The resulting crude reaction product was chromatographed on silica gel with MeOH/CHCl₃ (2:1) as eluent; yield 7c: 16%; mp > 139 °C decomp.; FAB-MS: m/e = 430.5 (100%), M+H⁺ = 430.5 calcd. for $C_{18}H_{31}N_5O_7$; HPLC (Nucleosil C_{18}): linear gradient from 100% A to 40% B in 25 min, t_R 20.80 min (80%); ¹H-NMR (DMSO): δ = 0.84 (d, 3H, $J = 6.0$ Hz, δ_2 CH₃ Leu); 0.88 (d, 3H, $J = 6.1$ Hz, δ_1 CH₃ Leu); 1.22 (t, 3H, $J = 7.0$ Hz, CL_3 -CH₂-O-); 1.36-1.68 (m, 7H, β CH₂ Leu, γ CH Leu, γ CH₂ Arg, β CH₂ Arg); 2.98-3.10 (m, 2H, δ CH₂ Arg); 3.71 (br s, 2H, 2 x Eps C-H); 3.84 (m, 1H, α CH Leu); 4.09 (q, 2H, J = 7.0 Hz, CH₃-CH₂-O-); 4.26 (m, 1H, α CH Arg); 7.46 (d, 1H, J = 7.5 Hz, NH Leu); 7.60-8.11 (br, HNC-NH₂); 8.75 (d, 1H, J = 9.1 Hz, NH Arg); 9.41 (br s, 1H, e NH Arg); ¹³C-NMR (DMSO): $\delta = 13.8$ (CH₃-CH₂-O-); 21.4 (δ CH₃ Leu); 23.0 (δ CH₃ Leu); 24.2 (γ CH Leu); 25.2 (γ CH₂ Arg); 29.6 (β CH₂ Arg); 38.9 (δ CH₂ Arg); 40.4 (β CH₂ Leu); 51.1, 51.7, 53.0, 53.4 (2 x Eps C-H, 2 x α CH); 61.5 (CH₃-CH₂-O-); 157.4 (HNC-NH₂); 165.0, 167.1, 170.0 (3 x CO); 174.6 (-COOH).

N-[[(2R,3R)-3-trans-(Ethoxycarbonyl)oxiran-2-yl]carbonyl]-leucyl-agmatine-1/2 H₂SO₄ (7d)

Compound **7d was synthesized** employing the procedure of **7c by** acylation of HCOOH-H-Leu-Agm-l/2 $H_2SO₄$ with freshly prepared 6a in presence of 1 eq. of NMM. The crude product was purified on a silica gel column (3 x 46 cm) using CHCl₃/MeOH/H₂O (15:10:1) as eluent; the combined fractions were concentrated to a small volume, washed with toluene (5x50 mL) and then evaporated under reduced pressure to semi-solid material; yield **7d: 77%;** FAB-MS: m/e = 386 (lOO%), M+H+ = 386.5 calcd. for $C_{17}H_{31}N_5O_5$; HPLC (Nucleosil C_{18}): linear gradient from 100% A to 40% B in 25 min, t_R 21.96 min (84%); ¹H-NMR (DMSO): $\delta = 0.83$ (d, 3H, J = 6.1 Hz, δ_2 CH₃ Leu); 0.87 (d, 3H, J = 6.1 Hz, δ_1 CH₃ Leu); 1.21 (t, 3H, J = 7.0 Hz, CH_3 -CH₂-O-); 1.38-1.50 (m, 6H, β CH₂ Leu, β CH₂ Agm, γ CH₂ Agm); 1.55 (m, 1H, γ CH Leu); 3.00-3.06 (br m, 4H, α CH₂ Agm, δ CH₂ Agm); 3.07 (d, 1H, J = 1.8 Hz, Eps C-H); 3.30 (d, 1H, J = 1.8 Hz, Eps C-H); 4.12 (q, 2H, J = 7.0 Hz, CH₃-CH₂-O-); 4.26 m, 1H, α CH Leu); 7.44-7.58 (br, 3H, HNC-NH2); 8.12 (t, lH, J = 5.5 Hz, NH Agm); 8.30 (br m, lH, E NH Agm); 8.70 (d, $1H, J = 8.2 Hz, NH Leu$).

N-[[(2R,3R)-3-trans-(Oxycarbonyl)oxiran-2-yl]carbonyl]-glycyl-proline dipotassium salt (8a)

To an ice-cold solution of **7a (204** mg, 0.65 nunol) in EtOH (3 mL) KOH (72.8 mg, 1.3 mmol) in EtOH (4 mL) was added. The reaction mixture was stirred for 2h at 0 $^{\circ}$ C and then diluted with ether. The precipitate was collected by filtration, washed extensively with ether, EtOH and again with ether; yield 8a: 226 mg (96%); mp > 178-179 °C decomp.; FAB-MS: m/e = 363 (40%), 131 (100%), M+2K⁺+H⁺ = 363.4, calcd. for C₁₇H₃₁N₃O₅; HPLC (Nucleosil C₁₈): linear gradient from 100% A to 40% B in 25 min, t_R 5.80 min (98%); ¹H-NMR (DMSO): δ = 1.60-1.72 (m, 2H, β CH₂ Pro); 1.92-2.05 (m, 2H, γ CH₂ Pro); 2.92 (d, 1H, $J = 2.2$ Hz, Eps C-H); 3.19 (d, 1H, $J = 2.2$ Hz, Eps C-H); 3.38-3.40 (m, 2H, δ CH₂ Pro); 3.68 (dd, 1H, $J =$ 5.0 Hz, J = 17.0 Hz, α_2 CH₂ Gly); 3.82 (dd, 1H, J = 5.6 Hz, J = 17.0 Hz, α_1 CH₂ Gly); 3.86 (dd, 1H, J = 8.2 Hz, J = 3.6 Hz, α CH Pro); 7.72 (t, 1H, J = 5.0 Hz, NH Gly); ¹³C-NMR (D₂O): δ = 25.1 (γ CH₂ Pro cis); 27.1 (γ CH₂ Pro trans); 32.3 (β CH₂ Pro trans); 34.5 (β CH₂ Pro cis); 44.3 (α CH₂ Gly); 49.6 (δ CH₂ Pro trans); 50.2 (δ CH₂ Pro cis); 55.8 (Eps C-H); 57.3 (Eps C-H); 64.7 (α CH Pro cis); 64.9 (α CH Pro trans); 172.8, 176.6 (2 x CO, Gly, Eps); 181.9, 182.5 (2 x COG-, Pro, Eps).

N-[[(2R,3R)-3-trans-(Oxycarbonyl)oxiran-2-yl]carbonyl]-leucyl-proline dipotassium salt (8b)

Compound **8b was** prepared from 7b (100 mg, 0.27 mmol) as described for 8a in 92% yield; mp > 223 "C decomp.; FAB-MS: m/e = 341.1 (30%), 183.4 (100%), M^+ = 341 .3 calcd. for C₁₅H₂₂N₂O₇; HPLC (Nucleosil C₈): linear gradient from 100% B to 100% C in 25 min, t_R 8.16 min (94%); ¹H-NMR (400 MHz, D₂O): δ = 0.94 (d, 3H, J = 6.0 Hz, δ_2 CH₃ Leu); 0.96 (d, 3H, J = 6.0 Hz, δ_1 CH₃ Leu); 1.64 (m, 2H, β CH₂ Leu); 1.67-1.76 (m, 1H, γ CH Leu); 1.85-1.95 (m, 1H, β_2 CH₂ Pro); 2.01 (m, 2H, γ CH₂ Pro); 2.19-2.27 (m, 1H, β_1 CH₂ Pro); 3.46 (d, 1H, J = 2.1 Hz, Eps C-H); 3.59 (d, 1H, J = 2.0 Hz, Eps C-H); 3.65 (m, lH, δ₂ CH₂ Pro); 3.82 (m, 1H, δ₁ CH₂ Pro); 4.27 (dd, 1H, J = 5.1 Hz, J = 8.6 Hz, α CH Leu); 4.59 (dd,

1H, J = 3.3 Hz, J = 10.1 Hz, α CH Pro cis); 4.70 (dd, 1H, J = 5.5 Hz, J = 8.7 Hz, α CH Pro trans); ¹³C-NMR (D₂O): $\delta = 23.4$ (δ_2 CH₃ Leu); 24.9 (γ CH₂ Pro cis); 25.3 (δ_1 CH₃ Leu); 27.2 (γ CH Leu); 27.3 (γ CH₂ Pro trans); 32.2 (β CH₂ Pro trans); 34.4 (β CH₂ Pro cis); 41.7 (β CH₂ Leu); 50.1 (δ CH₂ Pro cis); 50.4 (S CH, Pro trans); 53.0 (a CH Leu); 55.6 (Eps C-H); 56.9 (Eps C-H); 64.8 (a CH Pro cis); 64.9 (a CH Pro trans); 172.1, 174.5 (2 x CO, Leu, Eps); 176.6, 182.2 (2 x COO-, Pro, Eps).

N-[[(2R,3R)-3-trans-(Oxycarbonyl)oxiran-2-yl]carbonyl]-leucyl-arginine dipotassium salt (8c)

Following the procedure of **8a** compound 8c was obtained from 7c (58.7 mg, 0.137 mmol) in 60% yield after chromatographic purification by preparative HPLC using as eluent a linear gradient from 100% I to 40% IV in 25 min; the product was lyophilized twice from water; FAB-MS: $m/e = 402$ (90%), 93 (100%), M+H⁺ = 402.4 calcd. for C₁₆H₂₇N₅O₇; HPLC (Nucleosil C₁₈): linear gradient from 100% A to 40% C in 25 min, t_R 10.43 min (99%); ¹H-NMR (DMSO): $\delta = 0.80$ (d, 3H, J = 5.2 Hz, δ_2 CH₃ Leu); 0.85 (d, 3H, J $= 5.2$ Hz, δ_1 CH₃ Leu); 1.26-1.68 (m, 7H, β CH₂ Leu, γ CH Leu, γ CH₂ Arg, β CH₂ Arg); 2.84 (m, 1H, δ CH₂ Arg); 3.15 (m, 1H, δ CH₂ Arg); 3.09 (br s, 1H, Eps C-H); 3.34 (br s, 1H, Eps C-H); 3.79 (m, 1H, α CH Leu); 4.23 (m, 1H, α CH Arg); 7.36 (d, 1H, J = 5.2 Hz, NH Leu); 8.3-8.9 (br, 4H, HNC-NH₂, NH Arg); 9.6-9.8 (br, 1H, ε NH Arg); ¹³C-NMR (DMSO): $\delta = 21.1$ (δ CH₃ Leu); 23.0 (δ CH₃ Leu); 24.2 (γ CH Leu); 25.2 (γ CH₂ Arg); 29.2 (β CH₂ Arg); 38.9 (δ CH₂ Arg); 40.1 (β CH₂ Leu); 51.6 (α CH Leu); 52.4 (Eps C-H); 53.2 (a CH Arg); 54.4 (Eps C-H); 157.8 (HNC-NH2); 167.6, 170.3 (2 x CO, Leu, Eps); 174.5, 180.2 (2 x COO-, Arg, Eps).

N-[[(2R,3R)-3-trans-(Oxycarbonyl)oxiran-2-yl]carbonyl]-leucyl-agmatine potassium salt (8d)

Compound 8d was prepared from 7d (228 mg, 0.525 mmol) as described for 8a; the crude product was purified by preparative HPLC with a linear gradient from 100% II to 40% IV in 25 min; yield **8d:** 30%; mp 161-162 °C; FAB-MS: m/e = 358.3 (100%), M+H⁺ = 358.4 calcd. for C₁₅H₂₇N₅O₅; HPLC (Nucleosil) C_{18}): linear gradient from 100% A to 40% C in 25 min, t_R 17.33 min (99%); ¹H-NMR (DMSO): δ = 0.83 (d, 3H, J = 6.2 Hz, δ_2 CH₃ Leu); 0.87 (d, 3H, J = 6.2 Hz, δ_1 CH₃ Leu); 1.41-1.54 (m, 7H, β CH₂ Leu, γ CH Leu, β CH₂ Agm, γ CH₂ Agm); 3.01-3.09 (m, 4H, α CH₂ Agm, δ CH₂ Agm); 3.09 (d, 1H, J = 1.7 Hz, Eps C-H); 3.34 (d, 1H, J = 1.8 Hz, Eps C-H); 4.27 (m, 1H, α CH Leu); 7.42 (br s, 3H, HNC-NH₂); 8.04 (t, 1H, $J = 5.5$ Hz, NH Agm); 8.15 (d, 1H, $J = 8.5$ Hz, NH Leu); 8.72 (br s, 1H, ε NH Agm); ¹³C-NMR (DMSO): δ = 21.7 (δ CH₃ Leu); 22.7 (δ CH₃ Leu); 24.3 (γ CH Leu); 25.7, 26.2 (β CH₂, γ CH₂ Agm); 37.8 (δ CH₂ Agm); 40.3 (α CH₂ Agm); 40.7 (β CH₂ Leu); 51.4 (α CH Leu); 52.4 (Eps C-H); 54.6 (Eps C-H); 157.1 (HNC-NH₂); 167.6, 170.1 (2 x CO, Leu, Agm); 171.6 (COO⁻, Eps).

Reaction of 8a with 3-Methylbenzothiazole-2-thione

To an ice-cold solution of $7a$ (45.9 mg, 0.146 mmol) in either CH₂Cl₂ or CH₃CN (3 mL) 3methylbenzothiazole-2-thione (26.5 mg, 0.146 mmol) in the corresponding solvent (1.5 mL) and TFA (11.3 **pL, 0.146 mmol) were** added. No reaction was observed to occur within 3d at room temperature as judged by hplc even adding an additional equiv. of MBTT. After refluxing the solution for 3d, 7a was quantitatively converted into a complex mixaue of products. The solvent was then evaporated, the residue partitioned between water and EtOAc (3 x) to extract the MBTT and its corresponding oxide. The aquaeous layer was lyophilized and the residue chromatographed on a Sephadex RP 18 (25-40 μ m) column (2 x 135 cm) using 5% isopropanol (95% water) and 15% isopropanol (85% water) as eluent; flow rate: 204 mL/h. The main product could be identified by MS as the desulfurated fumaryl compound: $m/e = 299$ (20%), $M+H^+ = 299.3$ calcd. for $C_{13}H_{18}N_2O_6$.

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References

- $\mathbf{1}$. (a) Hanada, K.; Tamai, M.; Yamagishi, M.; Ohmura, S.; Sawada, J.; Tanaka, I.; *Agric. Biol. Chem.,* **1978,42,523-528. (b) Hanada,** K.; Tamai, M.; Ohmura, S.; Sawada, J.; Seki, T.; Tanaka, I.; *ibid.*, 529-536. (c) Hanada, K.; Tamai, M.; Morimoto, S.; Adachi, T.; Ohmura, S.; Sawada, S.; Tanaka, I.; *ibid, 537-541.*
- $2.$ (a) Tamai, M.; Hanada, K.; Adachi, T.; Oguma, K.; Kashiwagi, K.; Omura, S.; Ohzeki, M.; J. *Biochem.*, 1981, 90, 255-257. (b) Hanada, K.; Tamai, M.; Morimoto, S.; Adachi, T.; Oguma, K.; Ohmura, S.; Ohxeki, M.; in *Peptide Chemistry,* **Yonehara,** H. Ed.; Protein Research Foundation: Osaka, 1980; pp. 31-36. (c)Barrett, A. J.; Kembhavi, A. A.; Brown, M. A.; Kirschke, H.; Knight, C. G.; Tamai, M.; Hanada, K.; *Biochem. J., 1982,201,* 189-198. (d) Murata, M.; Miyashita, S.; Yokoo, C.; Tamai, M.; Hanada, K.; Hatayama, K.; Towatari, T.; Niiawa, T.; Katunuma, N.; *FEBSLeti., 1991,2&I, 307-310. (e)* Gour-Salin, B. J.; Lachance, P.; Plouffe, C.; Storer, A. C.; Menard, R.; J. Med. Chem., **1993**, 36, 720-725. (f) Giordano, C.; Calabretta, R.; Gallina, C.; Consalvi, V.; Scandurra, R.; Noya, F. C.; Franchini, C.; Eur. J. Med. Chem., 1993, 28, 917-926.
- 3. **(a) Yabe, Y.; Guillaume, D.; Rich, D. H.;** *J. Am. Chem. Soc.***, 1988, 110, 4043-4044. (b) Buttle, D. J.;** Murata, M.; Knight, C. G.; Barrett, A. I.; *Arch. Biochem. Biophys., 1992,299,377-380.*
- **4.** Towatari, T.; Niiwa, T.; Murata, M.; Yokoo, C.; Tamai, M.; Hanada, K.; Katunmna, N.; *FEBS Lett.,* **1991,280, 311-31s.**
- **5.** (a) van Tameleu, E. E.; J *Am. Chem. Sac,, 1951,* **73,3444-3448. (b)** Price, C. C.; Kirk, P. F.; J. *Am. Chem. Sot., 1953, 75,2396-2400. (c)* Ketcham, R; Shah, V. P.; J. Org. Chem., **1963,28,229-230.** (d) Chan, T. H.; Finkenbine, J. R.; *J. Am. Chem. Soc.*, 1972, 94, 2880-2882. (e) Calõ, V.; Lopez, L.; Marchese, L.; Pesce, G.; *J. Chem. Soc., Chem. Commun.*, 1975, 621-622.
- 6. **(a) Sato, R.; Okanuma, M.; Chida, S.; Ogawa, S.; Tetrahedron Lett., 1994, 35, 891-894. (b)** Bonnans-Plaisance, C.; Levesque, G.; Midrak, A.; *Eur. Polym, J.*, **1994**, 30, 239-244.
- **7.** (a) Tamai, M.; Yokoo, C.; Murata, M.; Oguma, K.; Sota, K.; Sato, E.; Kanaoka, Y.; Chem. *Phurm. Bull. Jpn.,* 1987,35, 1098-l 104. (b) Ohashi, I.; Harada, K.; *Bull. Chem. Sot. Jpn.,* **1967,40,2977-2979.**
- **8.** Tamai, M.; Ada&i, T.; Oguma, K.; Morimoto, S.; Hauada, K.; Ohmura, S.; Ohzeki, M.; *Agric. Biol. Chem., 1981,45,675-679.*
- **9.** (a) Mori, K.; Iwasawa, H.; *Tetruhedron,* **1980, 36, 87-90. (b)** Seebach, D.; Wasmuth, D.; *Helv. Chim. Acta, 1980, 63, 197-200.*
- 10. Appel, R.; Baumer, G.; Striiver, W.; *Chem. Ber.,* **1975,108,2680-2692.**
- 11. Korn, A.; Rudolph-Böhner, S.; Moroder, L.; *Tetrahedron*, 1994, *50*, 1717-1730.

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